

# Preparation of Multilayered Organic-Inorganic Nanohybrids via Layer-by-Layer Assembly

Kiyofumi Katagiri<sup>1,2</sup>, Shogo Iseya<sup>2</sup>, Atsunori Matsuda<sup>2</sup>, Kunihito Koumoto<sup>1</sup>

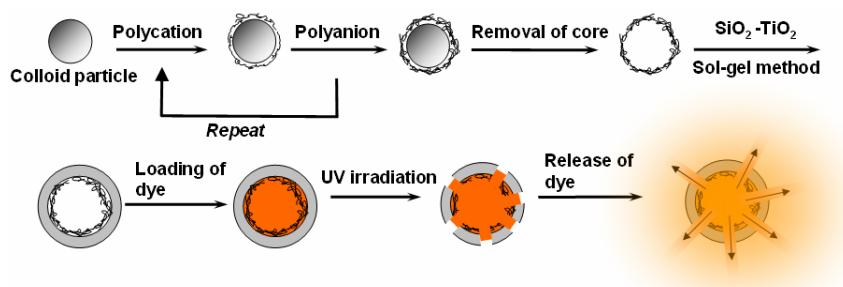
<sup>1</sup> Department of Applied Chemistry, Graduate School of Engineering, Nagoya University

<sup>2</sup> Department of Materials Science, Toyohashi University of Technology

E-mail: katagiri @ apchem.nagoya-u.ac.jp

**Keywords:** Layer-by-Layer Assembly, Sol-Gel Method, Organic-Inorganic Hybrid, Hollow Capsule

Layer-by-layer  
(LbL) assembly of oppositely charged polyelectrolytes is widely used as a versatile method for preparation of core-shell particles. Not only polyelectrolytes (PEs) but also a number

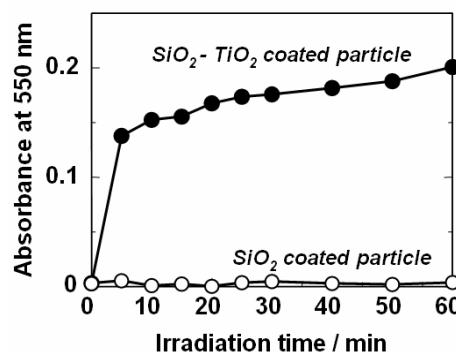


**Fig. 1.** Schematic representation of the formation procedure of organic-inorganic composite hollow capsule via layer-by-layer assembly technique with colloidal template.

of materials having multivalent charge, e.g., inorganic nanoparticles, are applicable for the layer-by-layer assembly technique [1]. Therefore, a novel class of organic-inorganic composite materials can be easily obtained by the incorporation of inorganic matters for LbL process. In this study, several organic-inorganic composite core-shell particles and hollow capsules were prepared via the layer-by-layer assembly technique. As an inorganic component,  $\text{TiO}_2$  was chosen as a photocatalyst. PEs- $\text{TiO}_2$  composite capsules are expected as control delivery vehicles (Fig. 1).

The UV responsive property of the hollow capsules was confirmed by release of dye molecules (Phenol Red) upon UV irradiation. In the case of the capsule formed with PEs and  $\text{SiO}_2$ , the dyes were not released from the capsules even after irradiation for 1 h. On the other hand, the dyes were quickly released upon UV irradiation for 5 min for the capsules formed with PEs and  $\text{SiO}_2\text{-TiO}_2$  (Fig. 2). It should be owing to formation of holes in the capsule shells by the photocatalytic decomposition of organic PE multilayers.

[1] F. Caruso, ed, *Colloids and Colloid Assemblies* (Wiley-VCH, Weinheim, 2004).



**Fig. 2.** Changes in the absorbance of phenol red released from the (PSS/PDDA)<sub>n</sub>/SiO<sub>2</sub> or SiO<sub>2</sub>-TiO<sub>2</sub> particles as a function of UV-irradiation time.

## Biographical Sketch



<b>Name:</b>	Kiyofumi Katagiri
<b>Born :</b>	Gifu, Japan in 1975
<b>Nationality :</b>	Japanese
<b>Affiliation :</b>	Department of Applied Chemistry, Graduate School of Engineering, Nagoya University Furo-cho, Chikusa-ku, Nagoya 464-8603, Japan Tel: +81-52-789-3330, Fax: +81-52-789-3201 E-mail: katagiri @ apchem.nagoya-u.ac.jp URL: <a href="http://www.apchem.nagoya-u.ac.jp/BS-6/seigyo6/">http://www.apchem.nagoya-u.ac.jp/BS-6/seigyo6/</a>
<b>Title :</b>	Assistant Professor
<b>Degree:</b>	Ph. D. (Nara Institute of Science and Technology, Japan)
<b>Award:</b>	Excellent Presentation Award, the Chemical Society of Japan (2006)
<b>Hobby:</b>	Trekking, Cycling, Hot spring, Soccer, etc.

Kiyofumi Katagiri is an assistant professor in the Department of Applied Chemistry, Graduate School of Engineering, Nagoya University. He received his BEng degree from Osaka Prefecture University (supervisors: Professors M. Tatsumisago and T. Minami), and his MEng and PhD degrees from Nara Institute of Science and Technology (supervisors: Professors K. Ariga and J. Kikuchi). He spent two years as a postdoctoral research fellow under the supervision of Professor Frank Caruso at the Centre for Nanoscience and Nanotechnology, Department of Chemical and Biomolecular Engineering, The University of Melbourne. He also spent one year as a research fellow in the Department of Materials Science, Toyohashi University of Technology (supervisors: Professor A. Matsuda). He was awarded a Research Fellowship for Young Scientists from the Japan Society for the Promotion of Science. His current research interests include biomimetic materials and organic-inorganic nanohybrid materials based on supramolecular chemistry, colloid and surface science, and sol-gel science and technology.